

OPTICAL ROTATORY DISPERSION STUDIES CXXI.¹ VERIFICATION OF

THE ABSOLUTE CONFIGURATION OF 2-METHYLCYCLOHEXANONE

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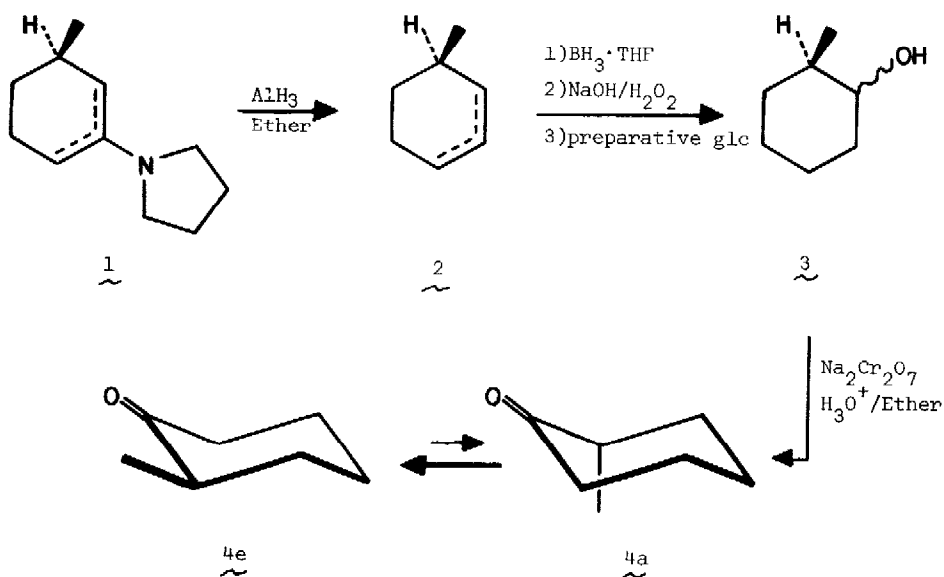
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Since its establishment as a rational basis for the Cotton effects of saturated ketones, the octant rule³ has been applied with great success to a wide variety of stereochemical and configurational problems.⁴ Because this rule correlates the immediate stereochemical environment of a dissymmetrically perturbed carbonyl group to the sign of its Cotton effect, it has become an important method in the determination of absolute configuration of saturated ketones.

Conformational analysis requires that 2-methylcyclohexanone exist chiefly as an equilibrium mixture of axial (4a) and equatorial (4e) chair conformers with the more stable 4e predominant. Considerable experimental evidence⁵ suggests that equatorial substituents make a negligible contribution to the Cotton effect, and are thus overwhelmed by the large rotatory contribution of the minor axial conformer. The negligible contribution of α -equatorial substituents has received theoretical support from a number of authors⁶ but has been seriously questioned by results of semi-empirical and *ab initio* calculations by Rauk, *et al.*⁷ Despite substantial scatter in computed rotatory strengths these authors assert that the equatorial isomer of (R)-2-methylcyclohexanone (4e) is predicted to exhibit an unexpectedly large positive rotation, and thus suggest that the previously assigned⁸ absolute configuration based on the octant rule is in error. Since this assertion raises serious fundamental questions, it seemed imperative to unambiguously verify this assignment.

Reduction of the pyrrolidine enamine (1) of (R)-3-methylcyclohexanone⁹ with AlH_3 ¹⁰ led to a mixture of (R)-3-, and (R)-4-methylcyclohexene (2)¹¹ which was not separated but subjected to hydroboration-oxidation¹² directly to provide a mixture of 2-, 3- and 4-methylcyclohexanols,¹¹ from which pure 2-methylcyclohexanol (3) could be separated by preparative glc (15% Carbowax 4000 on Chromosorb A, 130°). Two-phase oxidation¹³ afforded (R)(-)-2-methylcyclohexanone (4) (free of residual 3 by glc). The circular dichroism curve was negative, $[\theta]_{288\text{nm}}^{\text{max}} = -987^\circ$ ($c = 0.54\%$ in MeOH). Upon addition of base the cd slowly decreased to baseline, thus excluding the possibility of contamination of 4 by (R)-3-methylcyclohexanone.

The negative cd of 4 reconfirms the initial⁸ assignment of absolute configuration to this molecule as well as the absolute configurational assignments⁸ of the isomeric 2-methylcyclohexanols, all of which have been related to 4.



REFERENCES

1. For part CXX see J. W. Simek, D. L. Mattern and C. Djerassi, *Tetrahedron Lett.*, 3671 (1975)
2. Visiting scholar, 1975-1976, while on leave from the University of Rhode Island. Financial assistance provided by the National Institutes of Health (Grant No. GM 20276) is gratefully acknowledged.
3. W. Moffit, R. B. Woodward, A. Moscovitz, W. Klyne and C. Djerassi, *J. Amer. Chem. Soc.*, **83**, 4013 (1961).
4. For leading references see: Chapter 10 in C. Djerassi, "ORD: Applications to Organic Chemistry", McGraw-Hill, New York, 1960; C. Djerassi, *Pure & Applied Chem.*, **2**, 475 (1961) and P. Crabbé, "ORD and CD in Chemistry and Biochemistry", Academic Press, New York, 1974.
5. C. Beard, C. Djerassi, J. Sicher, F. Sipos and M. Tichy, *Tetrahedron*, **19**, 919 (1963), C. Djerassi and W. Klyne, *J. Chem. Soc.*, 2390 (1963).
6. T. D. Bouman and D. A. Lightner, *J. Amer. Chem. Soc.*, **98**, 3145 (1976) and references therein
7. A. Rauk, J. O. Jarvie, H. Ichimura and J. M. Barriell, *J. Amer. Chem. Soc.*, **97**, 5656 (1975).
8. C. Beard, C. Djerassi, T. Elliot and R. C. C. Tao, *J. Amer. Chem. Soc.*, **84**, 874 (1962).
9. E. J. Eisenbraun and S. M. McElvain, *J. Amer. Chem. Soc.*, **77**, 3383 (1955).
10. J. M. Coulter, J. W. Lewis and P. P. Lynch, *Tetrahedron*, **24**, 4489 (1968).
11. Identified by comparison (glc, ir and nmr) with authentic racemic compounds.
12. H. C. Brown, "Organic Syntheses via Boranes", John Wiley & Sons, New York, 1975.
13. H. C. Brown, C. P. Garg and K.-T. Liu, *J. Org. Chem.*, **36**, 387 (1971).